Supplemental Material

Enzymatic Synthesis and Flash Chromatography Separation of the Natural Phenylpropenoids, 1,3-Diferuloyl-sn-Glycerol and 1-Feruloyl-sn-Glycerol

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Table S1. CombiFlash Rf200i flash chromatography of ethyl ferulate standard at varying flow rates, 5 – 27 mL/min. See Figure S3 for experimental parameters.

<table>
<thead>
<tr>
<th>Flow Rate (mL/min)</th>
<th>Peak Base Width UV (CVa)</th>
<th>Peak Base Width ELSD (CV)</th>
<th>ELSD Peak Offsetb</th>
<th>Run Time (CV)</th>
<th>Run Time (min.)</th>
</tr>
</thead>
<tbody>
<tr>
<td>5.0</td>
<td>0.75</td>
<td>0.90</td>
<td>+1.00</td>
<td>20.0</td>
<td>19.2</td>
</tr>
<tr>
<td>10.0</td>
<td>0.88</td>
<td>0.98</td>
<td>+0.71</td>
<td>20.0</td>
<td>9.6</td>
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<tr>
<td>18.0</td>
<td>1.42</td>
<td>1.78</td>
<td>−0.15</td>
<td>20.0</td>
<td>5.3</td>
</tr>
<tr>
<td>27.0</td>
<td>1.78</td>
<td>2.58</td>
<td>−0.31</td>
<td>20.0</td>
<td>3.6</td>
</tr>
</tbody>
</table>

a The CombiFlash Rf200i measures run time in Column Volumes (CV) and minutes. The CV of the RediSep RF Gold 4-g Silica Gel Disposable column was 1 CV = 4.8 ml at 18 ml/min default flow rate.
b The difference between the start of the ELSD peak detection before (+) or after (−) the start of the UV peak detection.
Figure S1. Photograph of the CombiFlash Rf 200i flash chromatography system with a 4-g RediSep RF Gold Silica Gel Disposable Flash Chromatography Column (20 – 40 microns) as the “load column” containing the raw precipitate plumbed above a 24-g RediSep RF Gold Silica Gel Disposable Flash Chromatography Column (20 – 40 microns) “separation column.”
Figure S2. $^1$H NMR (500 MHz, $^d$-acetone) of FG and F$_2$G.
Figure S3. CombiFlash Rf200i flash chromatography of ethyl ferulate standard at varying flow rates, 5 – 27 ml/min. Ethyl ferulate standard, 10 mg, dissolved in 250 l acetone was 1-ml disposable syringe injected onto a RediSep RF Gold 4-g Silica Gel Disposable column (1 Column Volume, CV, = 4.8 ml at 18 ml/min default flow rate) and aspirated to dryness under vacuum for 30 min. The column was developed as follows: liquid injection mode of 3.0 CV equilibration with 100 % (v:v) hexane (solvent A), followed by 0 – 100 % gradient (blue line) with acetone (solvent B, right-blue axis) for 10 CV, followed by 5 CV of 100 % acetone, followed by 5 CV of 100 % hexane. The signal was monitored by UV (325 nm, left-red axis, purple trace) and ELSD (left-green axis, green trace).
**Figure S4.** Photograph of F:G and FG purified by flash chromatography of the crude precipitate (3 g) collected from the lipase-catalyzed transesterification of Enova Oil with ethyl ferulate.
Figure S5. Chemical structures of the commercial UVB absorbing ingredient, Octinoxate, and the commercial UVA absorbing ingredient, Avobenzone.
Figure S6. Absorbance extinction coefficient ($\varepsilon$, M$^{-1}$ cm$^{-1}$) of FG and F$_2$G in ethanol solutions determined as the slope of the linear regression of absorbance at $\lambda_{max}$, $n = 3$ trials.
Figure S7. Absorbance extinction coefficient ($\varepsilon$, M$^{-1}$ cm$^{-1}$) of FG and F$_2$G in acetonitrile solutions determined as the slope of the linear regression of absorbance at $\lambda_{max}$, $n = 3$ trials.